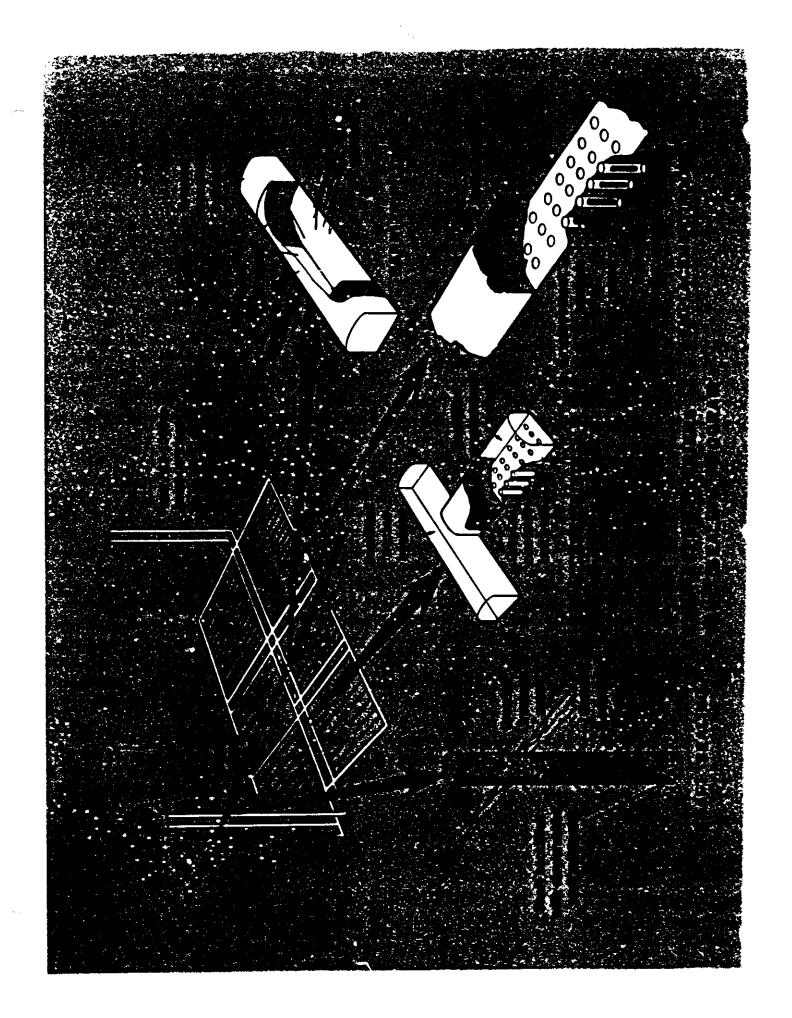
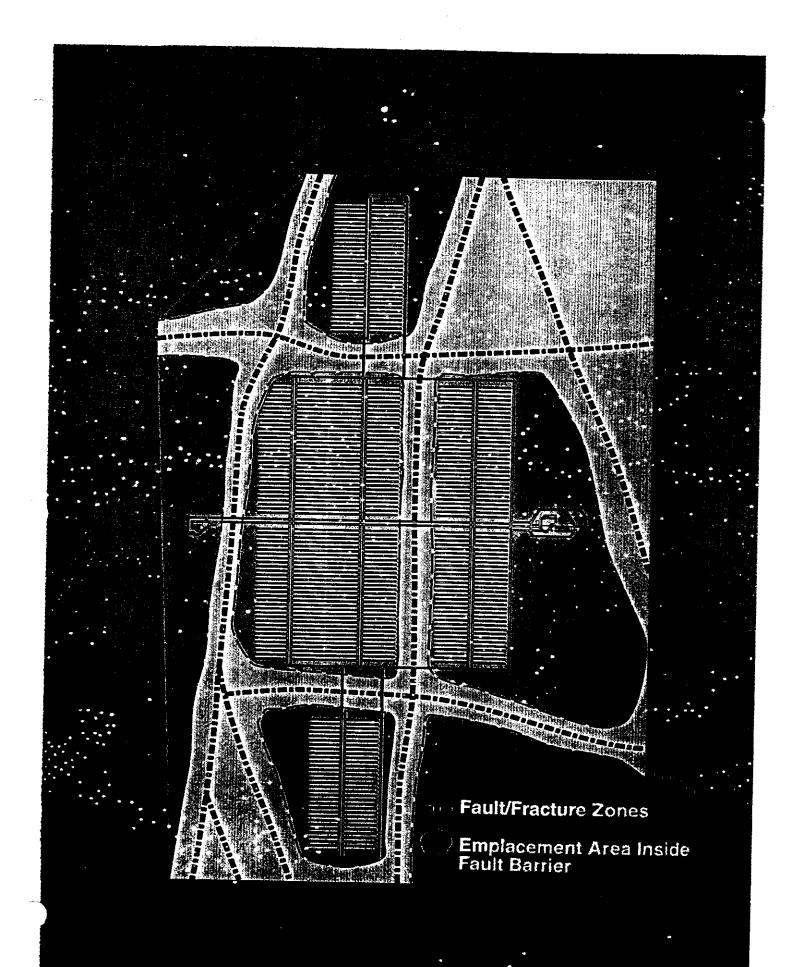
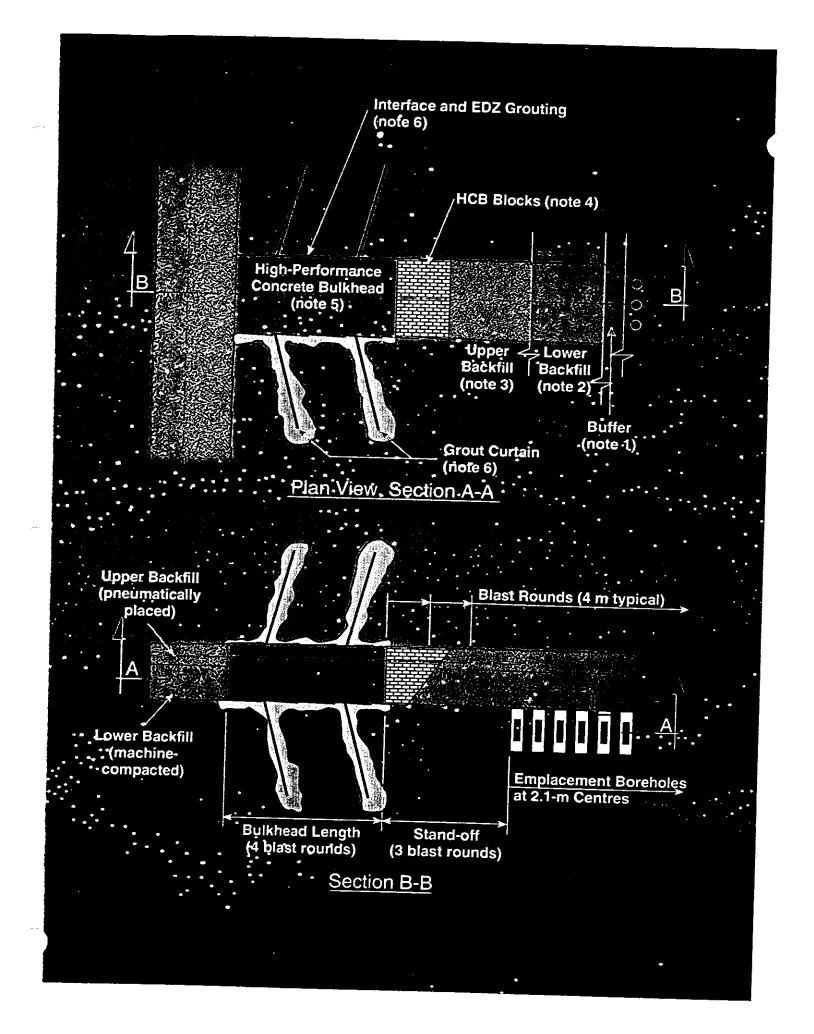
♦ SEALING MATERIALS SEAL DESIGN ♦ SEAL PERFORMANCE

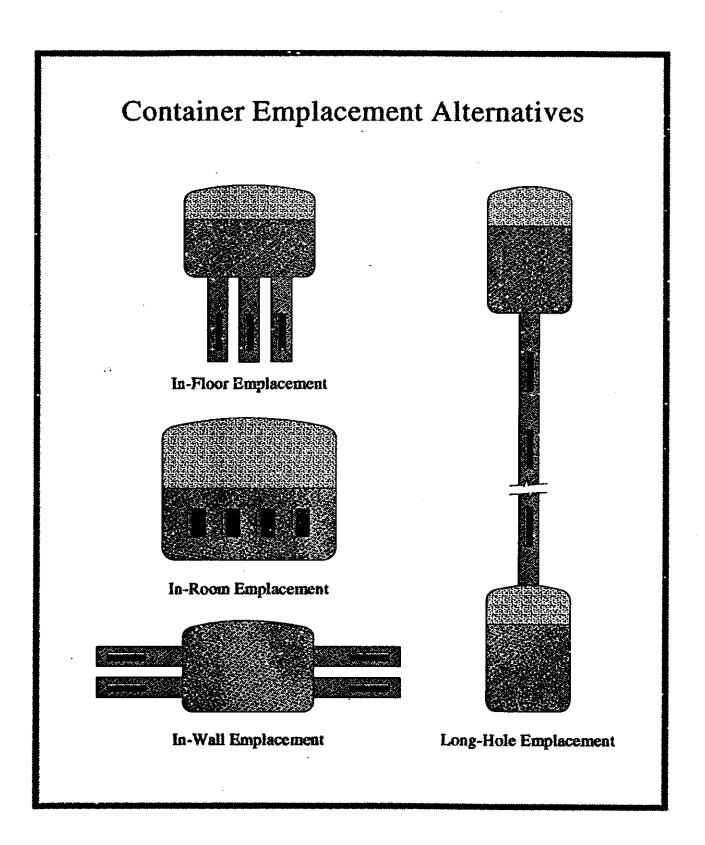
SEALING STRATEGIES

- Minimize Water Movement Around the Waste Container
- Decrease Hydraulic Conductivity in the Vault
- Seal Hydraulically Critical points in the Vault
- Enhance Sorption of Radionuclides and Chemically Condition the Groundwater









SEALING SYSTEM REQUIREMENTS

'n

Cool			
Seal	Engineering Objective	Performance	Approaches Used
		Requirement	in EIS Vault Model
Buffer	Clay Dry Density >1.24 Mg·m ⁻³	No convection	No convection;
			always in transport
	Hyd. Conductivity, (k) $<10^{-11}$ m s ⁻¹		path
		•	
Backfill	$k < 10^{-10} \text{ m} \cdot \text{s}^{-1}$	No/minimal	In transport path,
		convection	for rooms below FZ
Bulkheads,			
Shaft Seals			
- bentonite	Density >2 Mg⋅m ⁻³	No convection	Evaluated in
	$k < 10^{-11} \text{ m/s}^{-1}$		detailed model, not
		i.	in vault or
- concrete	Provide physical support to	Minimal	geosphere model
	backfill	alteration of	
		buffer/backfill	
EDZ			
- grouts	Use where $k > 10^{-7} \text{ m} \cdot \text{s}^{-1}$;	EDZ should	Evaluated in
-	reduction of k by 10 to 100	not be a flow	detailed model, not
		path; e.g.,	in vault or
- rock	Optimize execution to prevent	keyed-in seals	
	Optimize excavation to prevent	reyeu-iii seals	geosphere model
	connected permeability		

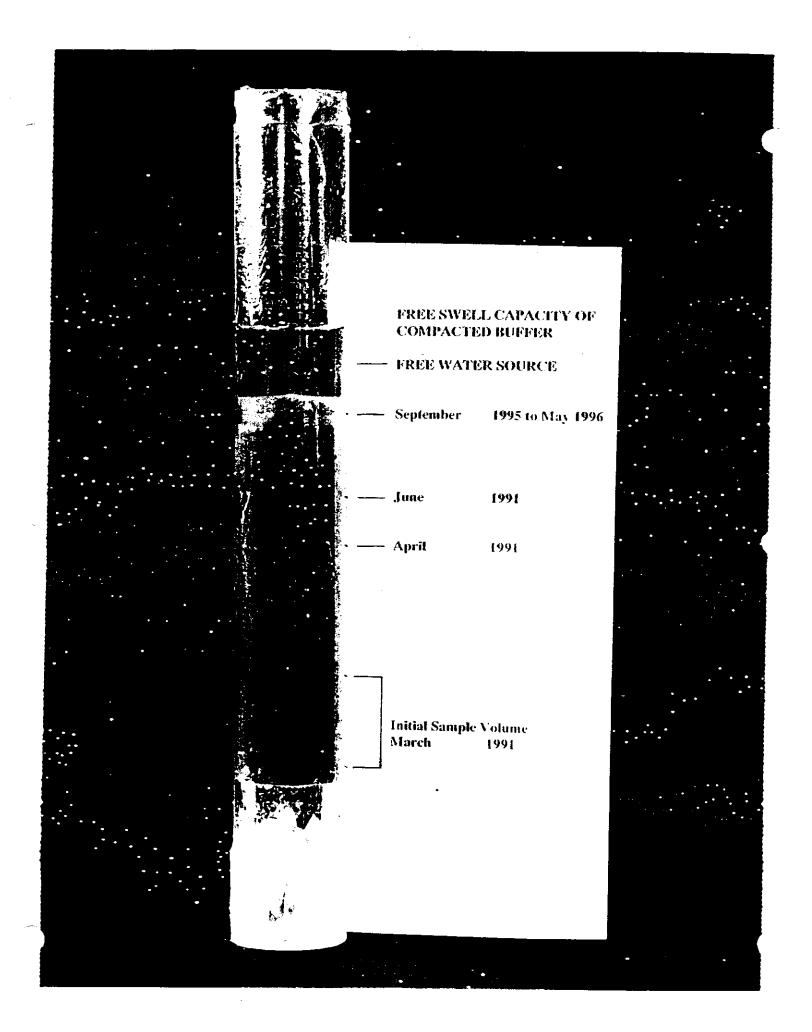
VAULT SEALING MATERIALS

Clay-Based Materials

- Low Hydraulic Conductivity
- Swelling and Extrusion
- Sorption
- Neutral pH
- Emplacement Options
 - In-situ Compaction
 - Precompacted Blocks
 - (Aggregate Addition
- Availability

Cement-Based Materials

- Low Hydraulic Conductivity
- High Strength
- Engineering Material Many Options



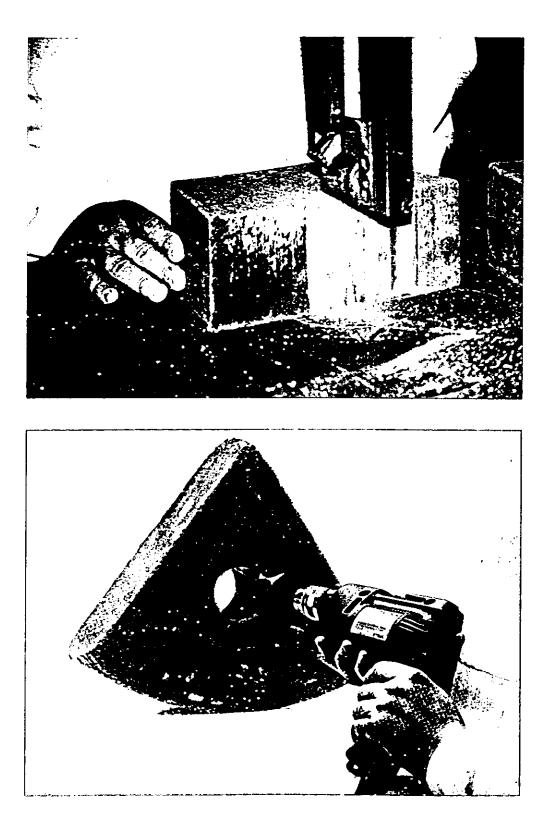


FIGURE 4-15: Large Precompacted Blocks of the Reference Buffer Material Being Sawn in a Band Saw (top) and Being Augered (bottom)

BUFFER AND BACKFILL PERFORMANCE

• Smectite \rightarrow Illite

Swelling

Gas Generation and Transport

Cements

Radiation

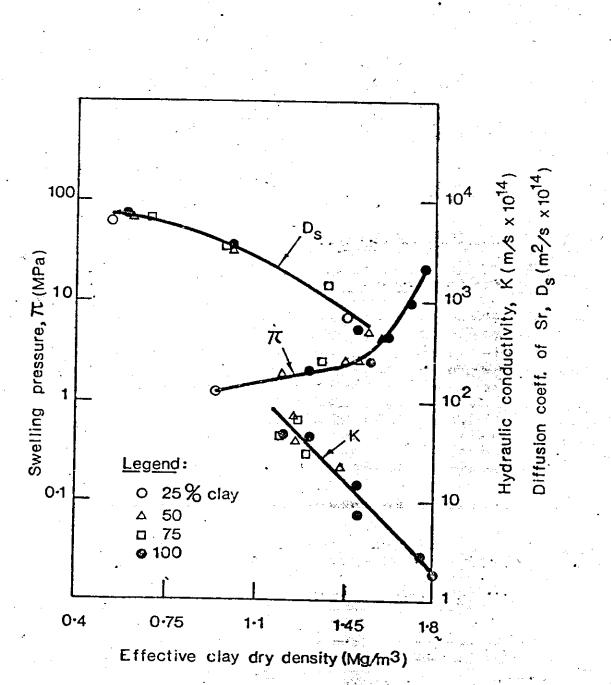
Microbial Activity

Colloids

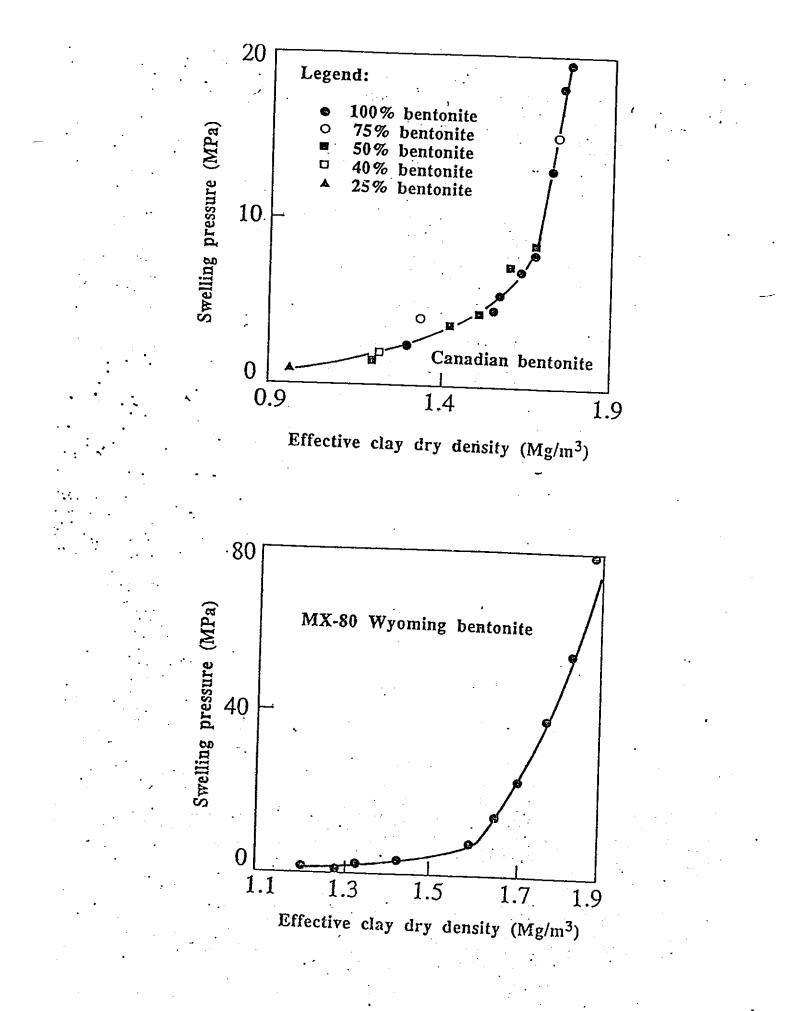
Mineralogical composition and related chemistry of Avonlea bentonite

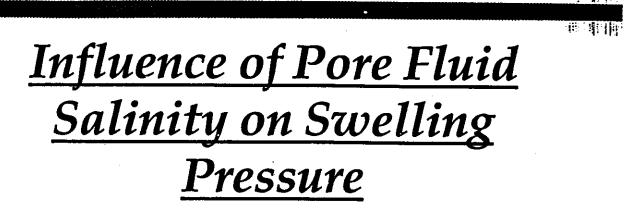
	%	
Montmorillo	nite 79	
Illite	10	
Quartz	5	
Feldspar	3	
Gypsum	2	
Carbonate	1	
Organic Mat	ter 0.3	
_		

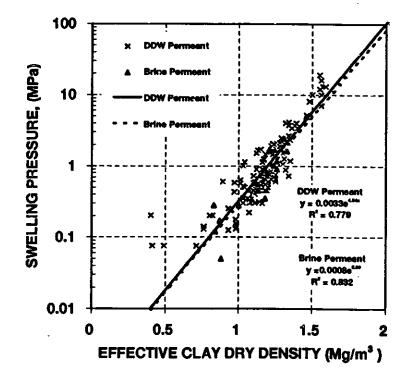
SSA= 630 x 10^3 m²/kg; CEC= 82 cmol_c/kg; exchangeable cations in cmol_c/kg: Na⁺= 47, Ca²⁺= 40, Mg²⁺= 7, K⁺= 0.7.



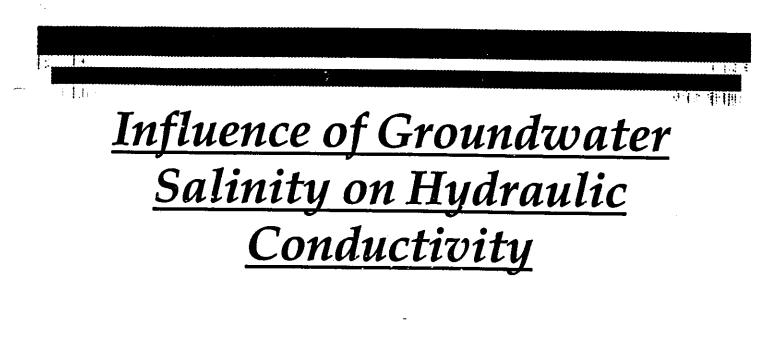
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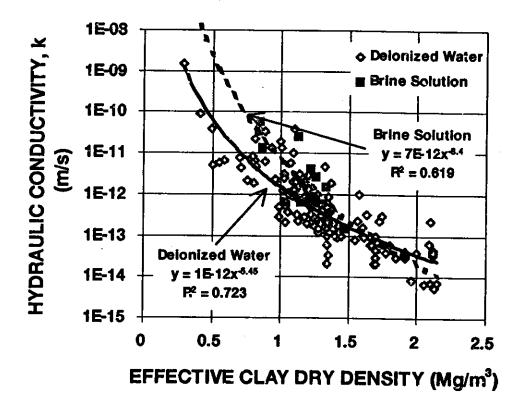


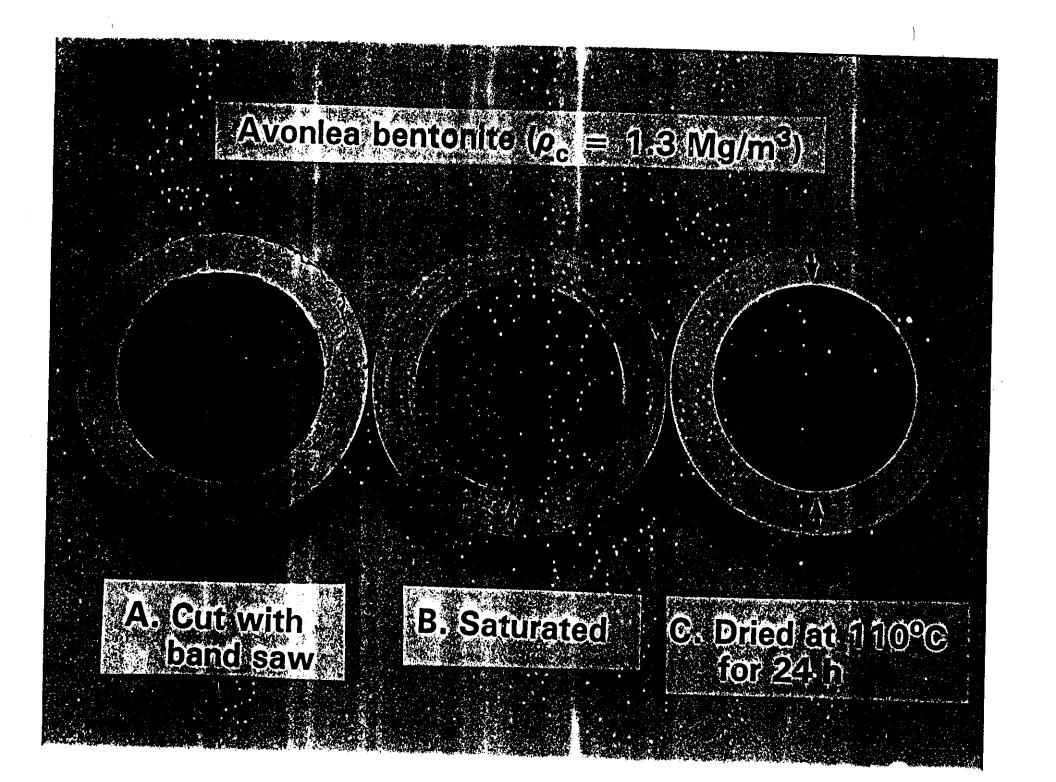


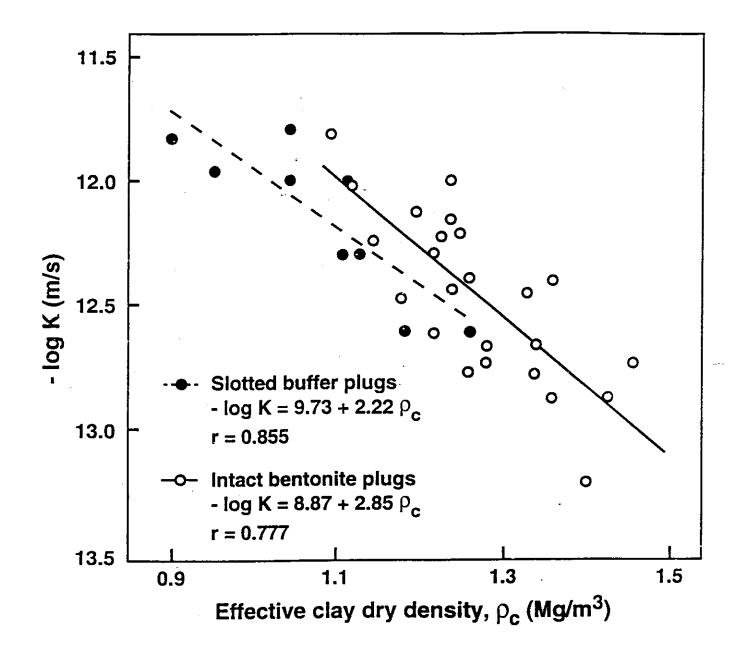


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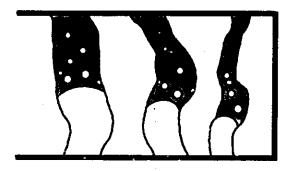


<u>Gas Breakthrough</u> Resistance in Bentonite

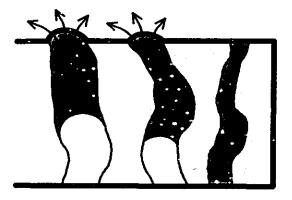
lechanisms that may create athways through porous media

- Diffusion
- Capillarity
- Pathway Dilatancy
- Tensile Fracturing
- 'ossible gas pressure conditions
- gas pressure < pore water (PWP)</p>
- gas pressure >PWP but < total soil pressure
- gas pressure > total soil pressure

GAS FLOW MECHANISMS



DIFFUSION DISSOLUTION OF GASES IN THE WATER PHASE



2-PHASE FLOW

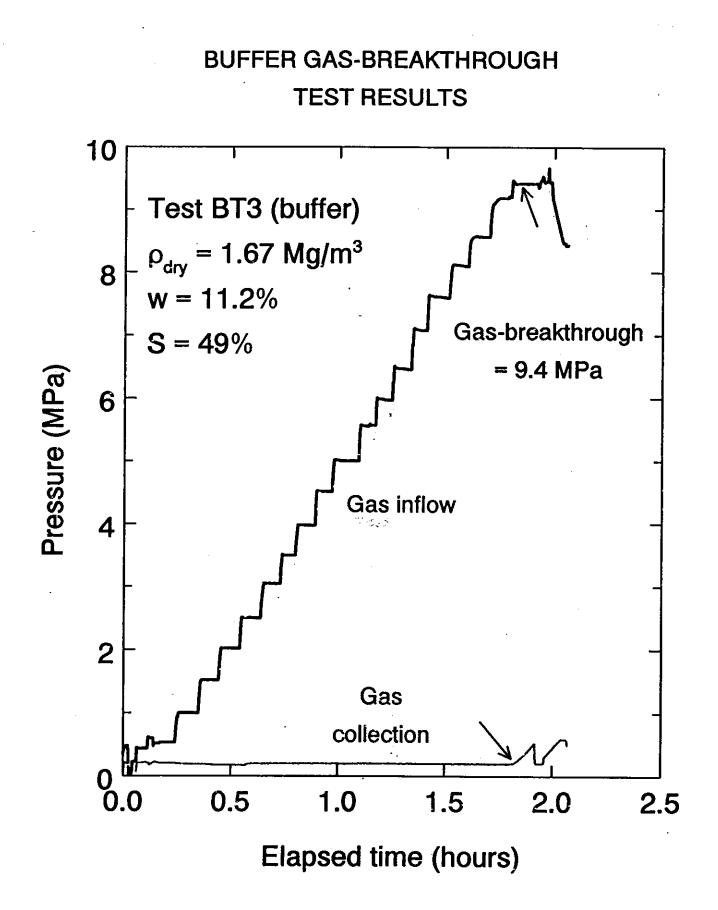
WATER IS PUSHED THROUGH SOME PORES BY INVADING GAS

PORE DILATION DEFORMATION OF SOIL FABRIC CREATING LARGER PORES TO ACCOMODATE GAS FLOW



FISSURING

CREATION OF NEW PORES TO ACCOMODATE GAS FLOW



Diffusion in Buffer and Backfill

 $k < 10^{-10}$ m/s - diffusion dominant

D, Total Intrinsic Diffusion Coefficient from

 $J = -D (\partial c / \partial x); \quad D = D_o \tau \varepsilon$

D_a, Apparent Diffusion Coefficient from

$$\frac{\partial c}{\partial t} = D_{a} \left(\frac{\partial^{2} c}{\partial x^{2}} \right); D_{a} = \frac{D_{o} \tau \varepsilon}{\varepsilon + \rho K_{d}} = \frac{D}{r}$$

 $r = Capacity Factor (\epsilon + \rho K_d)$

- D and r from Laboratory Experiments
 - Literature
 - Expert Judgement

Diffusion Coefficients, D_a, in Buffer

Diffusant	D _a (µm²/s)	Breakthrough time [*] (years)
ľ	100	20
Cs⁺	1	2000
Pu	0.01	200 000

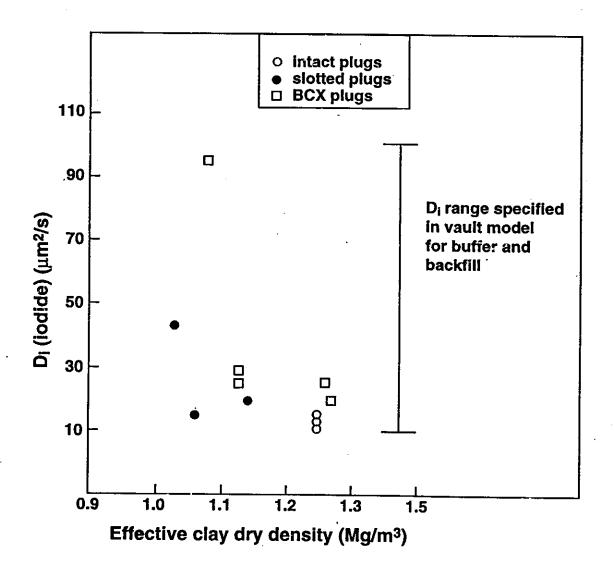
Approximate time required for $c/c_o = 0.5$ at the buffer/rock interface; buffer thickness = 0.25 m

Diffusion Coefficients for Large Molecules (MW 354 to 3000)

<0.001 µm²/s

(Eriksen and Jacobsson, KBS TR-84-05)

Total intrinsic diffusion coefficients, D_i, for I⁻ in intact and defected bentonite plugs





Cement-based Materials

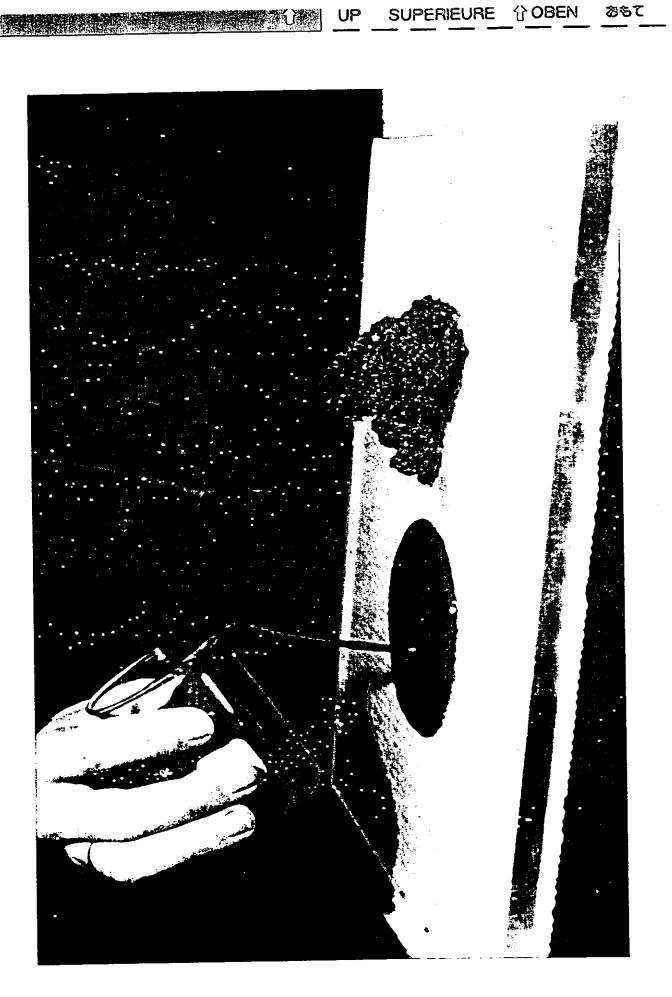
- For high-level waste disposal, generally restricted to grouting, shaft seal and construction applications (e.g., bulkheads, floors)
- Low pH concretes have been developed that are more compatible with clay buffers and backfills

CNFWMP Reference Grout

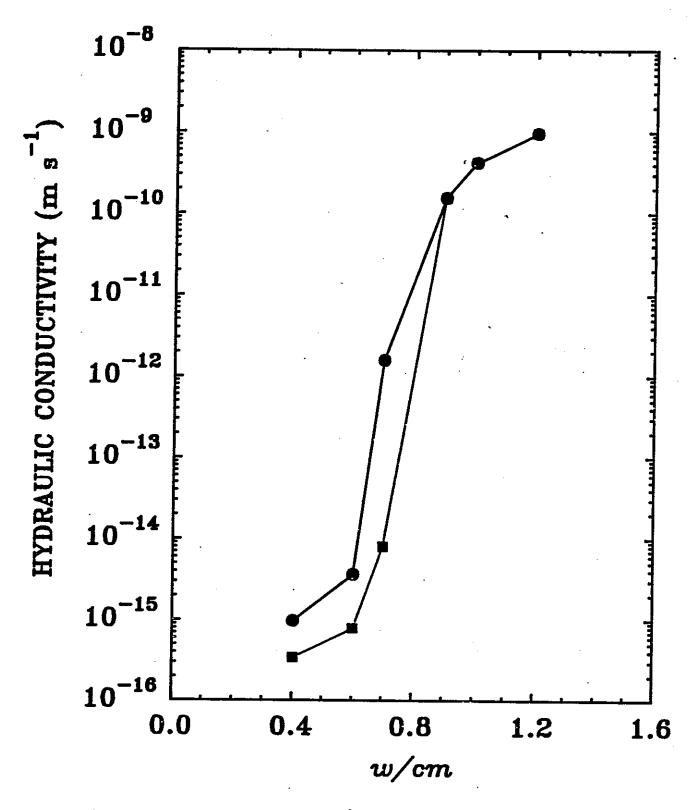
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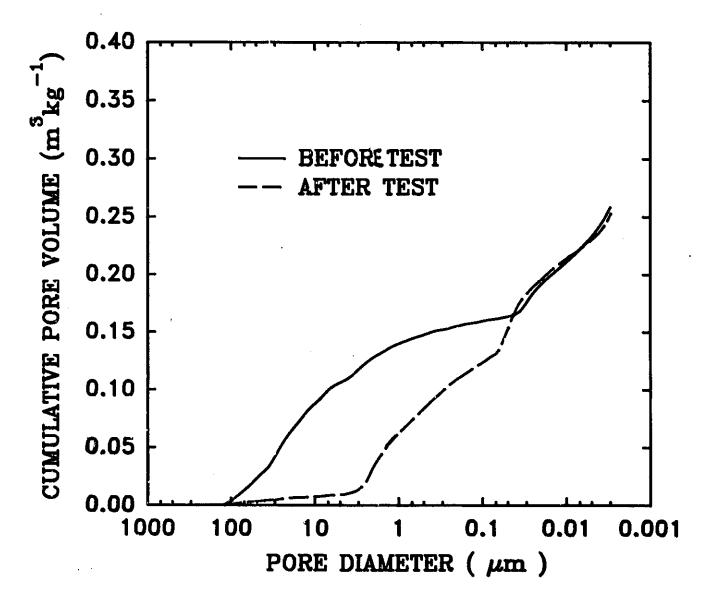
Cement Type :	Canadian Type 50 Reground to 600 m ² /Kg (Blaine)
Pozzolan :	Silica Fume (10% of total dry mass)
Superplasticizer :	Na-sulphonated naphthalene formaldehyde condensate (liquid)
Mass ratio of : water to (cement+pozzolan)	0.35 to 0.6
Superplasticizer : content	Varies with desired viscosity. Typical values 0.75 to 1.5 percent dry mass ratio superplasticizer to



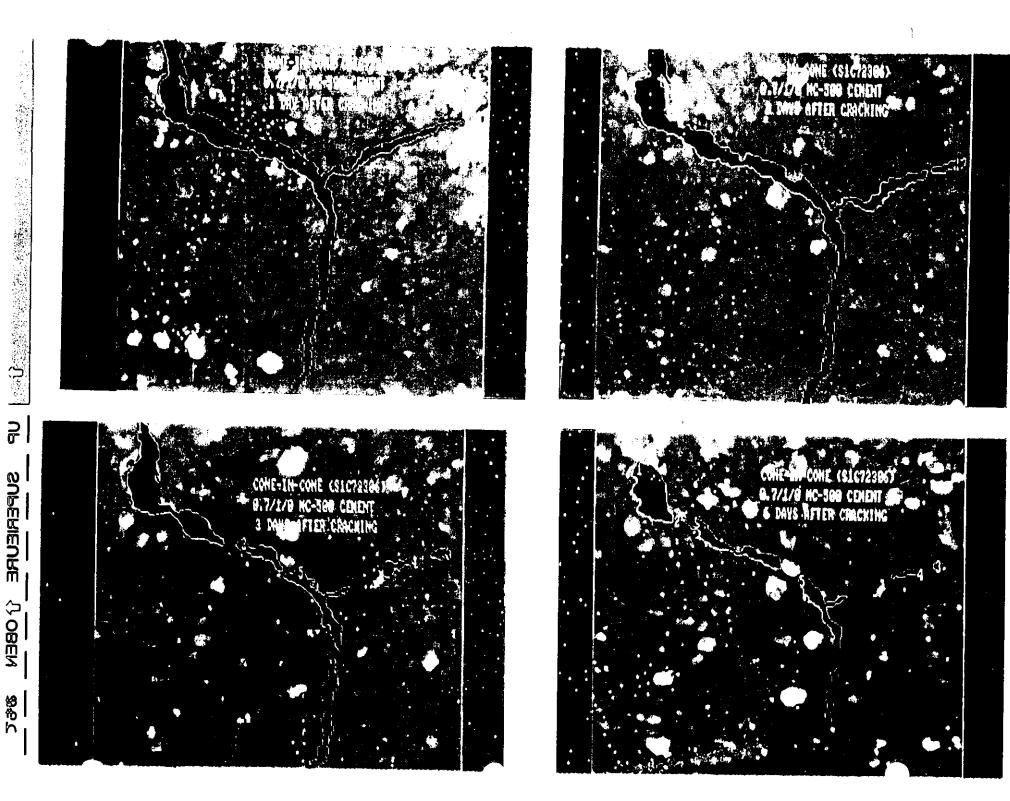
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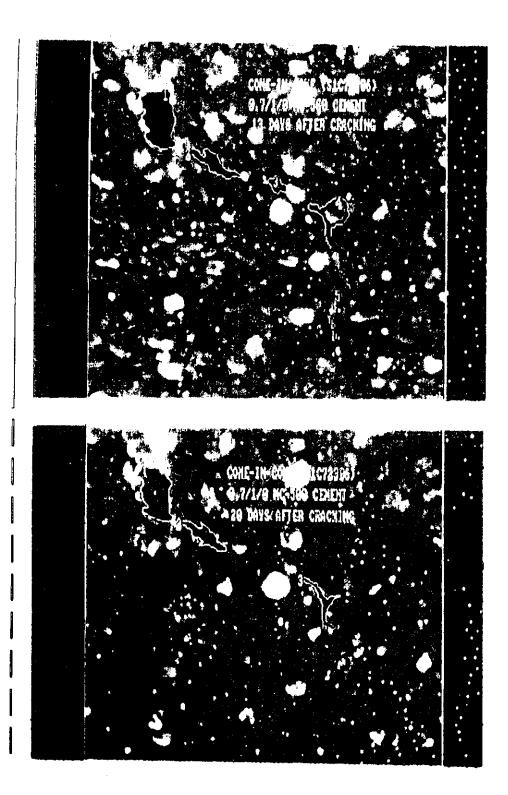


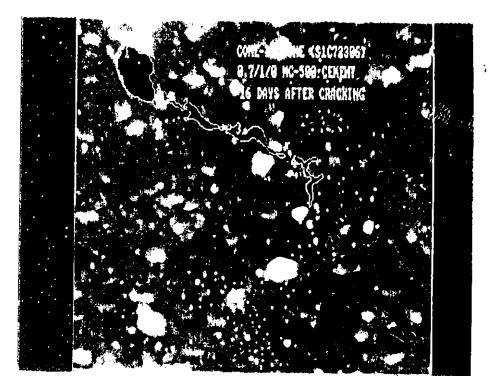
EFFECT OF WATER/CEMENTITIOUS MATERIAL RATIOS ON THE HYDRAULIC CONDUCTIVITY OF GROUTS

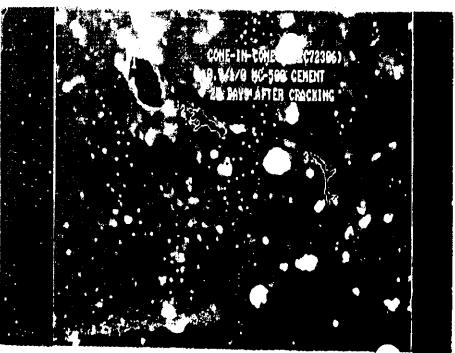


CHANGE IN PORE-SIZE DISTRIBUTION OF CEMENT-BASED GROUT (W/CM=0.4, TWO PARTICLE SIZES : $\Phi = 1.18$ mm AND $\Phi = 0.30$ mm) COMPACTED AT $\rho = 1.6$ Mg m⁻³.









Properties	LHHPC	Normal
	(w/cm 0.47)	(w/cm 0.56)
Fresh concrete		
Slump (mm)	160	170
Air Content (%)	2.75	2.75
Maximum temperature rise during hydration (°C)	15	~ 45
Maximum temperature during hydration (°C) -	37	~ 65
Hardened concrete		
Density (kg/m ³)	2424	2168
Hydraulic conductivity (m/s)	10^{-13} to 10^{-12}	10^{-11} to 10^{-12}
рН	9.65	~ 12.5
Total porosity - MIP technique (ml/g)	0.0580	n/a
Drying shrinkage - 90 days in air (με)	863	n/a
Drying shrinkage - 7 days in water and 83 days in air (µɛ)	348	n/a
Drying shrinkage - 21 days in water and 69 days in air (µɛ)	171	n/a
Drying shrinkage - 90 days in water (µɛ)	-50	n/a
Compressive strength - 28 days, 23°C (MPa)	86	29
Young's modulus - 28 days, at 40% of ultimate stress (GPa)	36.26	21.89
Poisson's ratio - 28 days, at 40% of ultimate stress	0.114	0.087

Properties of fresh and hardened LHHPC and normal concrete.

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